

**Liliana Dobrzańska\*‡ and  
 Gareth O. Lloyd**

Department of Chemistry, University of Stellenbosch, Private Bag X1, Matieland 7602, South Africa

‡ Permanent address: Faculty of Chemistry, Nicolaus Copernicus University, Toruń, Poland.

Correspondence e-mail: lianger@sun.ac.za

**Key indicators**

Single-crystal X-ray study  
 T = 100 K  
 Mean  $\sigma(C-C)$  = 0.008 Å  
 R factor = 0.069  
 wR factor = 0.136  
 Data-to-parameter ratio = 16.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

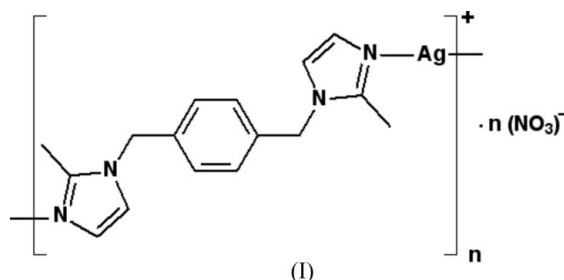
**catena-Poly[[silver(I)- $\mu$ -1,4-bis(2-methyl-1H-imidazol-1-ylmethyl)benzene- $\kappa^2 N^3:N^{3'}$ ] nitrate]**

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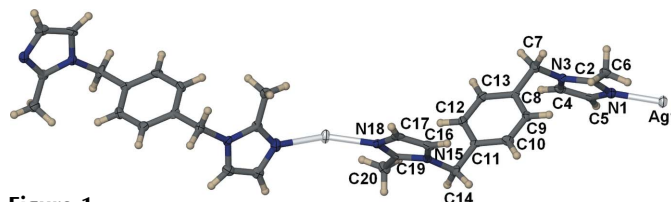
In the title complex,  $\{[Ag(C_{16}H_{18}N_4)]NO_3\}_n$ , the cationic part forms infinite chains where the  $Ag^I$  ions are coordinated in a slightly bent fashion by two N-donor atoms of imidazole rings of two distinct ligand molecules. The nitrate anions interact weakly with the silver cations to form infinite strands running in the *c*-axis direction, approximately perpendicular to the cationic chains. The resulting layers are stabilized by offset  $\pi$ - $\pi$  interactions between benzene rings. Furthermore, a set of C-H...O and  $\pi$ - $\pi$  interactions (imidazole rings) between adjacent layers results in the formation of a three-dimensional assembly.

**Comment**

During the course of our ongoing studies of metal complexes with flexible ditopic ligands (Dobrzańska, 2005; Dobrzańska, Lloyd *et al.*, 2005; Dobrzańska, Raubenheimer & Barbour, 2005; Dobrzańska *et al.*, 2006; Dobrzańska & Lloyd, 2006), we have isolated the title compound, (I), which consists of infinite cationic chains (Fig. 1) and nitrate counter-ions.



Bidentate 1,4-bis(2-methylimidazol-1-ylmethyl)benzene adopts the *trans* configuration with respect to the plane of the aromatic spacer and acts as a linkage between  $Ag^I$  ions [ $Ag \cdots Ag$  separation of 14.847 (4) Å]. The deviation from linearity about the Ag ion is significant [ $N-Ag-N = 155.78$  (17)°] and results from weak interactions of Ag ions with nitrate ions that function as bidentate (O1, O3) as well as bridging (O1, O3) units. In this manner, four Ag-O ‘quasi’ bonds are formed *via* three nitrate anions (two of which are



**Figure 1**  
 Fragment of an infinite cationic chain in the crystal structure of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 70% probability level.

generated by symmetry) and complete the overall environment about each Ag ion with the following Ag $\cdots$ O distances: bidentate Ag $\cdots$ O1 = 2.652 (4) Å and Ag $\cdots$ O3 = 3.042 (6) Å; bridging Ag1<sup>i</sup> $\cdots$ O1–N21–O3 $\cdots$ Ag1<sup>ii</sup> = 2.991 (4) and 2.995 (5) Å [symmetry codes: (i)  $x, \frac{1}{2} - y, \frac{1}{2} + z$ ; (ii)  $x, \frac{1}{2} - y, z - \frac{1}{2}$ ].

The presence of interactions between Ag and nitrate ions leads to the formation of infinite strands {Ag $\cdots$ NO<sub>3</sub>}<sub>n</sub> along the [001] direction, with an Ag $\cdots$ Ag distance of 4.785 (1) Å. Furthermore,  $\pi$ – $\pi$  stacking between benzene rings (centroid–centroid distance = 3.847 Å) results in the formation of an undulating layer (Fig. 2) which is schematically presented in Fig. 3.

The final three-dimensional assembly is stabilized by C–H $\cdots$ O hydrogen bonds with C $\cdots$ O distances in the range 3.39–3.89 Å (Table 1) and offset  $\pi$ – $\pi$  (imidazole rings) interactions (centroid–centroid distances 3.707 and 4.246 Å) between neighboring layers (Fig. 4).

## Experimental

A methanol solution of AgNO<sub>3</sub> was added to a methanol solution of 1,4-bis(2-methylimidazol-1-ylmethyl)benzene in a 1:1 molar ratio. Colorless crystals suitable for single-crystal X-ray diffraction were obtained by slow evaporation.

### Crystal data

[Ag(C <sub>16</sub> H <sub>18</sub> N <sub>4</sub> )]NO <sub>3</sub>	Z = 4
$M_r = 436.22$	$D_x = 1.790 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 14.885 (4) \text{ \AA}$	$\mu = 1.27 \text{ mm}^{-1}$
$b = 14.459 (4) \text{ \AA}$	$T = 100 (2) \text{ K}$
$c = 7.693 (2) \text{ \AA}$	Block, colorless
$\beta = 102.115 (4)^\circ$	0.12 $\times$ 0.10 $\times$ 0.09 mm
$V = 1618.7 (8) \text{ \AA}^3$	

### Data collection

Bruker APEX CCD area-detector diffractometer	9697 measured reflections
$\omega$ scans	3702 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1997)	2785 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.862, T_{\max} = 0.894$	$R_{\text{int}} = 0.056$
	$\theta_{\text{max}} = 28.3^\circ$

### Refinement

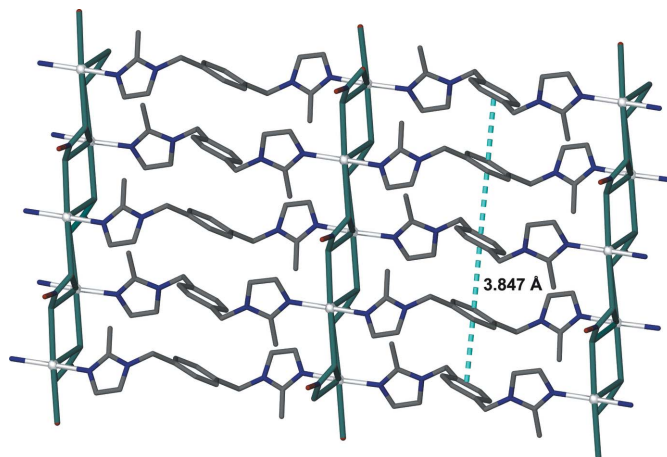
Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.05P)^2 + 3.538P]$
$R[F^2 > 2\sigma(F^2)] = 0.069$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.136$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.11$	$\Delta\rho_{\text{max}} = 1.34 \text{ e \AA}^{-3}$
3702 reflections	$\Delta\rho_{\text{min}} = -1.34 \text{ e \AA}^{-3}$
228 parameters	
H-atom parameters constrained	

**Table 1**

C–H $\cdots$ O2 interactions stabilizing the overall three-dimensional assembly of (I) (Å, °).

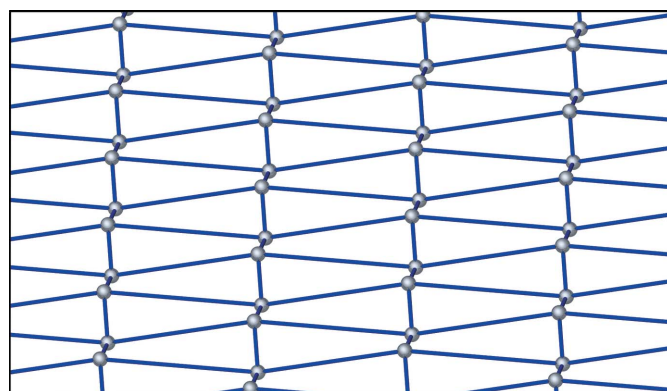
D–H $\cdots$ A	D–H	H $\cdots$ A	D $\cdots$ A	D–H $\cdots$ A
C10–H10 $\cdots$ O2 <sup>i</sup>	0.95	2.64	3.387 (7)	136
C14–H14b $\cdots$ O2 <sup>i</sup>	0.99	3.17	3.886 (8)	130
C13–H13 $\cdots$ O2 <sup>ii</sup>	0.95	2.66	3.404 (8)	135
C7–H7b $\cdots$ O2 <sup>ii</sup>	0.99	3.07	3.778 (8)	129

Symmetry codes: (i)  $-x, -y + 1, -z$ ; (ii)  $-x, y - \frac{1}{2}, -z + \frac{1}{2}$ .



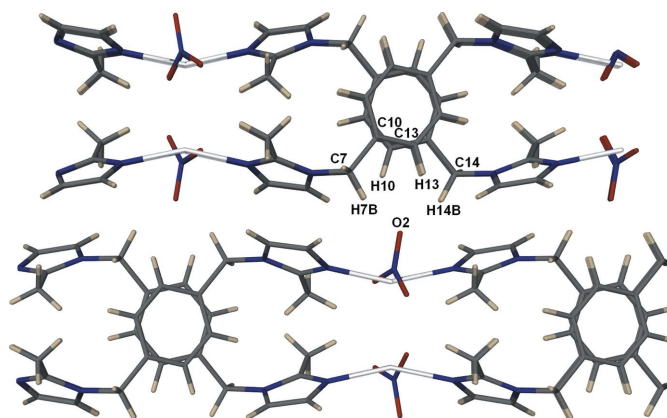
**Figure 2**

Capped-stick representation, showing layers formed by {Ag $\cdots$ NO<sub>3</sub>}<sub>n</sub> interactions (green strands) and  $\pi$ – $\pi$  stacking (blue dashed lines) in (I).



**Figure 3**

Schematic projection of the two-dimensional network.



**Figure 4**

A packing diagram of (I), viewed along [001], with labeled atoms participating in the stabilization of the overall three-dimensional assembly.

H atoms were positioned geometrically, with C–H = 0.95, 0.99 and 0.98 Å for aromatic, methylene and methyl H, respectively, and constrained to ride on their parent atoms, with  $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{C})$ , where  $x = 1.5$  for methyl H and  $x = 1.2$  for all other H. The highest peak and deepest hole are located 0.93 and 0.99 Å, respectively, from atom Ag1.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve

structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Barbour, 2001; Atwood & Barbour, 2003); software used to prepare material for publication: *X-SEED*.

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### References

- Atwood, J. L. & Barbour, L. J. (2003). *Cryst. Growth Des.* **3**, 3–8.
- Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
- Bruker (2001). *SMART*. Version 5.625. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2002). *SAINT*. Version 6.36a. Bruker AXS Inc., Madison, Wisconsin, USA.
- Dobrzańska, L. (2005). *Acta Cryst.* **E61**, o4113–o4115.
- Dobrzańska, L. & Lloyd, G. O. (2006). *Acta Cryst.* **E62**, o1205–o1207.
- Dobrzańska, L., Lloyd, G. O., Raubenheimer, H. G. & Barbour, L. J. (2005). *J. Am. Chem. Soc.* **127**, 13134–13135.
- Dobrzańska, L., Lloyd, G. O., Raubenheimer, H. G. & Barbour, L. J. (2006). *J. Am. Chem. Soc.* **128**, 698–699.
- Dobrzańska, L., Raubenheimer, H. G. & Barbour, L. J. (2005). *Chem. Commun.* pp. 5050–5052.
- Sheldrick, G. M. (1997). *SHELXS97*, *SHELXL97* and *SADABS* (Version 2.05). University of Göttingen, Germany.